Chapter

01

Halogen Derivatives





NEET-FLASHBACK



Q.1 Consider the reactions:

(i)
$$(CH_3)_2 CH - CH_2 Br \xrightarrow{C_2 H_5 OH}$$

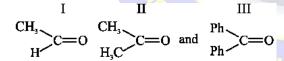
 $(CH_3)_2 CH - CH_2OC_2H_5 + HBr$

(ii)
$$(CH_3)_2 CH - CH_2Br \xrightarrow{C_2H_5O^-}$$

$$(CH_3)_2 CH - CH_2 OC_2 H_5 + Br^-$$

The mechanisms of reactions (i) and (ii) are respectively: [AIPMT-2011]

- (1) $S_N 2$ and $S_N 2$
- (2) S_N2 and S_N1
- (3) $S_N 1$ and $S_N 2$
- (4) S_N1 and S_N1
- Q.2 The order of reactivity of phenyl magnesium bromide (PhMgBr) with the following compounds: [AIPMT-2011]



- (1) | > | | > | |
- (2) | > | > | | > | | |
- (3) | || > || > 1
- (4) | | > | > | | |
- **Q.3** Which one is a nucleophilic substitution reaction among the following?

[AIPMT(Pre)-2011]

- (1) $CH_3CHO + HCN \rightarrow CH_3CH(OH)CN$
- (2) $CH_3-CH=CH_2+H_2O\longrightarrow CH_3-CH-CH_3$ OH
- (3) $RCHO + R'MgX \longrightarrow R-CH-R$ OH

- CH₃
 CH₃—CH₂—CH—CH₂Br + NH₃

 CH₃
 CH₄
 CH₅—CH—CH₂NH₂
- Q.4 Which of the following acids does not exhibit optical isomerism? [AIPMT(Pre)-2012]
 - (1) Tartaric acid
- (2) Maleic acid
- (3) α-amino acids
- (4) Lactic acid
- Q.5 In the following sequence of reactions $CH_3 Br \xrightarrow{KCN} A \xrightarrow{H_3O^+} B \xrightarrow{LiAlH_4} C$, the end product (C) is: [AIPMT(Pre)-2012]
 - (1) Ethyl alcohol
- (2) Acetone
- (3) Methane
- (4) Acetaldehyde
- Q.6 Which of the following compounds will undergo racemisation when solution of KOH hydrolyses?

[AIPMT-2014]

(ii) CH₃CH₂CH₂Cl

- (1) (ii) and (iv)
- (2) (iii) and (iv)
- (3) (i) and (iv)
- (4) (i) and (ii)
- Q.7 Two possible stereo-structures of CH₃CHOHCOOH, which are optically active, are called: [AIPMT-2015]

- (1) Mesomers
- (2) Diastereomers
- (3) Atropisomers
- (4) Enantiomers
- **Q.8** In a S_N1 reaction on chiral centres, there is:

[AIPMT-2015]

- (1) 100% inversion
- (2) 100% racemisation
- (3) Inversion more than retention leading to partial racemisation
- (4) 100% retention
- **Q.9** Which of the following reaction(s) can be used for the preparation of alkyl halides?

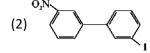
[AIPMT-2015]

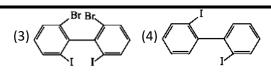
- (I) CH₃CH₂OH+HCl anh. ZnCl₂
- (II) CH₃CH₂OH+HCl→
- (III) $(CH_3)_3 COH + HCI \rightarrow$
- (IV) $(CH_3)_2$ CHOH+HCI anh. $ZnCl_2$
- (1) (III) and (IV) only (2) (I), (III) and (IV) only
- (3) (I) and (II) only
- (4) (IV) only
- Q.10 Consider the reaction

CH₃CH₂CH₂Br+NaCN → CH₃CH₂CH₂CN+NaBr

This reaction will be the fastest in: [NEET-2016]

- (1) N, N'-dimethylformamide (DMF)
- (2) Water
- (3) Ethanol
- (4) Methanol
- Q.11 Which of the following biphenyls is optically active: [NEET-2016]





- Q.12 An example of a sigma bonded organometallic compound is: [NEET-2016]
 - (1) Cobaltocene
- (2) Ruthenocene
- (3) Grignard's
- (4) Ferrocene
- Q.13 Which of the following will react faster through S_N1 mechanism? [NEET-2017]
 - (1) CH₃CH₂Cl
- (2) $H_2C = CH CH_2CI$
- (3) CI
- (4) $CH_2 = CHCI$
- Q.14 Of the following alcohols, the one that would react fastest with conc. HCl and anhydrous ZnCl₂ is: [NEET-2017]
 - (1) 2-methylpropanol
 - (2) Butan-1-ol
 - (3) Butan-2-ol
 - (4) 2-methylpropan-2-ol
- Q.15 Elimination reaction of 2-Bromo-pentane to form pent-2-ene is [NEET-2020]
 - (A) β-Elimination reaction
 - (B) Follows Zaitsev rule
 - (C) Dehydrohalogenation reaction
 - (D) Dehydration reaction
 - (1) (A), (C), (D)
 - (2) (B), (C), (D)
 - (3) (A), (B), (D)
 - (4) (A), (B), (C)

- Q.16 Which of the following will NOT undergo S_N1 reaction with OH₋? [2020 Covid Re-NEET]
 - (1) (CH₃)₃CCl (2)
 - (3) CH_2CI $(4) CH_2 = CH CH_2CI$
- Q.17 The intermediate compound 'X' in the following chemical reaction is: [NEET-2021]
 - $CH_3 + CrO_2Cl_2 \xrightarrow{CS_2} X \xrightarrow{H_3O^+} C$
 - (1) CH(OCoCH₂)₂ (2) CH(OCOCH₃)₂
 - $(3) \qquad CH \qquad Cl \qquad (4) \qquad CH \qquad H$
- Q.18 The major product formed in dehydrohalogenation reaction of 2 bromo pentane is pent 2 ene. This product formation is based on? [NEET 2021]
 - (1) Huckel's Rule (2) Saytzeff's Rule
 - (3) Hund's Rule (4) Holfmann Rule
- Q.19 The correct sequence of bond enthalpy of
 - 'C—X' bond is [NEET 2021]
 - (1) $CH_3 CI > CH_3 F > CH_3 Br > CH_3 I$
 - (2) $CH_3 f < CH_3 CI < CH_3 Br < CH_3 I$
 - (3) $CH_3 f > CH_3 CI > CH_3 Br > CH_3 I$
 - (4) $CH_3 f < CH_3 CI > CH_3 Br > CH_3 I$
- Q.20 The incorrect statement regarding chirality is

[NEET - 2022]

- (1) $S_N 1$ reaction yields 1:1 mixture of both enantiomers
- (2) The product obtained by $S_N 2$ reaction of haloalkane having chirality at the reactive site shows inversion of configuration
- (3) Enantiomers are superimposable mirror images of each other
- (4) A racemic mixture shows zero optical rotation.
- Q.21 The given compound

is an example of

[NEET - 2023]

- (1) Aryl halide
- (2) Allylic halide
- (3) Vinylic halide
- (4) Benzylic halide

ANSWER KEY

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Que.	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15
Ans.	1	2	4	2	1	3	4	3	2	1	3	3	2	4	4
Que.	16	17	18	19	20	21									
Ans.	2	1	2	3	3	2									



