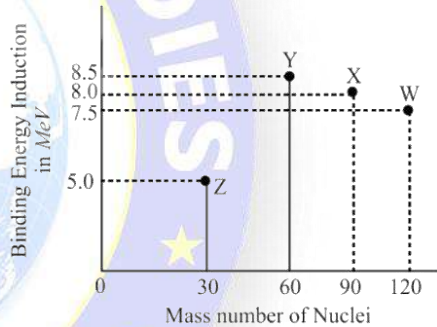


## DPP : PHYSICS – NUCLEAR PHYSICS

1. Which of the following pair of nuclei are isotones?  
 (a)  ${}^9_4\text{Be}, {}^{10}_5\text{B}$                       (b)  ${}^4_2\text{He}, {}^4_2\text{He}$   
 (c)  ${}^{17}_8\text{O}, {}^{17}_9\text{F}$                       (d)  ${}^7_3\text{Li}, {}^7_4\text{Be}$
2. Two protons are kept at a separation of 10 nm. Let  $F_n$  and  $F_e$  be the nuclear force and the electromagnetic force between them-  
 (a)  $F_e = F$   
 (b)  $F \gg F_n$   
 (c)  $F_e \ll F_n$   
 (d)  $F_e$  and  $F_n$  differ only slightly
3. The nuclei  ${}^{13}_6\text{A}$  and  ${}^{14}_7\text{B}$  can be described as  
 (a) Isotones  
 (b) Isobars  
 (c) Isotopes of carbon  
 (d) Isotopes of nitrogen
4. The nuclear radius as compared to the atomic radius is of the order  
 (a)  $10^{-3}$                       (b)  $10^{-5}$   
 (c)  $10^{-7}$                       (d)  $10^{-9}$
5. When two nuclei of mass X and Y respectively fuse to form a nucleus of mass m along with the liberation of some energy, then  
 (a)  $X + Y > m$                       (b)  $X - Y = m$   
 (c)  $X + Y = m$                       (d)  $X + Y < m$
6. Radius of  ${}^4_2\text{He}$  nucleus is 3 Fermi. The radius of  ${}^{32}_{16}\text{S}$  nucleus will be  
 (a) 6 Fermi                      (b) 4 Fermi  
 (c) 5 Fermi                      (d) 8 Fermi
7. Order of magnitude of density of uranium nucleus is [ $m_p = 1.67 \times 10^{-27} \text{ kg}$ ].  
 (a)  $10^{20} \text{ kg/m}^3$                       (b)  $10^{17} \text{ kg/m}^3$   
 (c)  $10^{14} \text{ kg/m}^3$                       (d)  $10^{11} \text{ kg/m}^3$
8. Two nuclei have mass number in ratio of 1 : 4. The ratio of their nuclear densities is:  
 (a) 1 : 4                      (b) 1 : 2  
 (c) 1 : 64                      (d) 1 : 1
9. The binding energy of nucleus is a measure of its  
 (a) Mass                      (b) Stability  
 (c) Charge                      (d) Momentum
10. Binding energy per nucleon vs mass number curve for nuclei is shown in figure. W, X, Y and Z are four nuclei indicated on the curve. The process that would release energy is  


Point	Mass Number	Binding Energy per Nucleon (MeV)
Z	30	5.0
Y	60	8.5
X	90	8.0
W	120	7.5

 (a)  $Y \rightarrow Z$                       (b)  $W \rightarrow X + Z$   
 (c)  $W \rightarrow 2Y$                       (d)  $X \rightarrow Y + Z$
11. The masses of neutron and proton are 1.0087 amu and 1.0073 amu respectively. If a helium nucleus (alpha particles) of mass 4.0015 amu is formed by combining neutrons and protons. The binding energy of the helium nucleus will be (1 amu 931MeV)  
 (a) 24.8MeV                      (b) 28.4MeV  
 (c) 14.2MeV                      (d) 42.8MeV
12. The binding energy of  $\alpha$ -particle  ${}^4_2\text{He}$  is 7.047 MeV per nucleon and the binding energy of deuteron  ${}^2_1\text{H}$  is 1.112 MeV per nucleon. Then in the fusion reaction  ${}^2_1\text{H} + {}^2_1\text{H} \rightarrow {}^4_2\text{He} + Q$ , the energy Q released is

- (a) 23.74MeV      (b) 32.82MeV  
(c) 11.9MeV      (d) 4.94MeV

**13.** The binding energy per nucleon for  ${}_{6}^{12}\text{C}$  nucleus is (Nuclear mass of  ${}_{6}^{12}\text{C} = 12.00000$  amu.

Mass of hydrogen nucleus = 1.007825 amu,  
Mass of neutron = 1.008665 amu.)

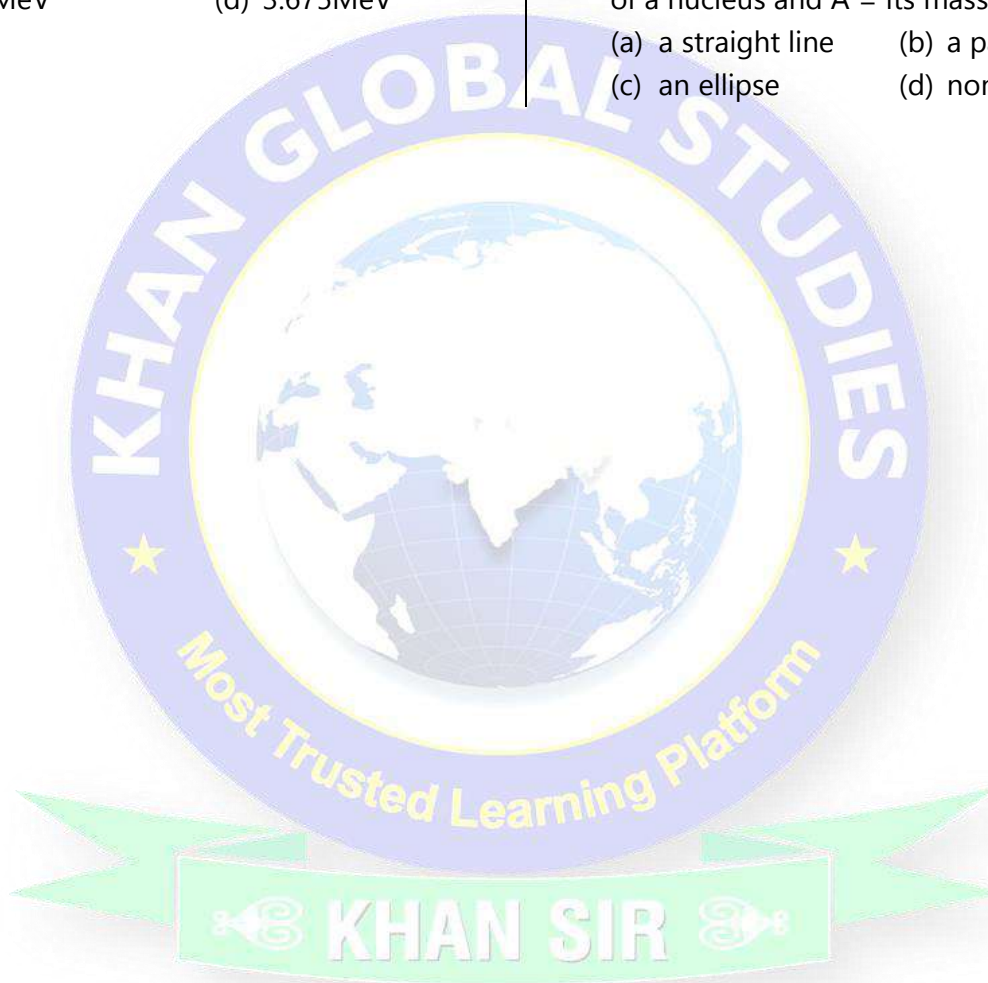
- (a) 2.675MeV      (b) 7.675MeV  
(c) 0MeV      (d) 3.675MeV

**14.** Consider the nuclear reaction  $X^{200} \rightarrow A^{110} + B^{90}$ . If the binding energy per nucleon for X, A and B is 7.4 MeV, 8.2MeV and 8.2MeV respectively, then the amount of the energy released is

- (a) 200 MeV      (b) 160 MeV  
(c) 110 MeV      (d) 90 MeV

**15.** The graph of  $\ln(R/R_0)$  versus  $\ln A$  (R = radius of a nucleus and A = its mass number) is-

- (a) a straight line      (b) a parabola  
(c) an ellipse      (d) none of these



**ANSWER KEY**

1. (a)	4. (b)	7. (b)	10. (c)	13. (b)
2. (b)	5. (a)	8. (d)	11. (b)	14. (b)
3. (a)	6. (a)	9. (b)	12. (a)	15. (a)

